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 $C_{18}H_{38} + 27O_2 \rightarrow 18CO_2 + 18H_2O$   
test tube 1 and the nail in  
test tube 5 give your answer to three significant figures 4  
**4 7 organic chemistry chemistry benton park school** Feb 06  
2021 web 2 pour the liquid hydrocarbon into the funnel 3 time  
how long it takes for all of the liquid hydrocarbon to run out of  
the funnel 4 repeat the experiment for the other liquid  
hydrocarbons 1 1 give the name of apparatus a in figure 1 1 mark  
1 2 name the apparatus that could be used to measure 40cm<sup>3</sup> of  
liquid hydrocarbon 1 mark

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also emitted by automobile exhausts burning of coal etc 3 flame  
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a carbon dioxide

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triple bonds to form between carbon atoms and even for  
structures such as rings to form saturated hydrocarbons have as  
many hydrogen

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have

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oxygen c carbon d both hydrogen and carbon answer d  
clarification these organic compounds are made up of only carbon  
and hydrogen and

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complete the equation for the cracking of decane by writing the  
correct formula and the name of the third product c 10 h 22 c 5 h  
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the formula

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2020 web these hydrocarbons are called alkane alkene and alkyne respectively carbons can be found linking in a closed cycle containing a benzene ring these cyclic hydrocarbons are called aromatic alkanes are called saturated hydrocarbon whereas all the other types are unsaturated chemical and physical properties differ for each type of hydrocarbon

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stoves and bunsen burners can be either propane c 3 h 8 or butane c 4 h 10 combustion of hydrocarbons all hydrocarbons burn cleanly to form water and carbon dioxide in the presence of

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Sep 13 2021 web the crude oil is pumped into the bottom of the column where it is so hot that the majority of the hydrocarbons turn from a liquid to gas the chains that are too long are drained out this is the residue that we call bitumen the gaseous hydrocarbon chains travel up the column since gas rises when a chain comes to a bit in the column where

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Mar 15 2019 web 10 jan 2023 aromatic hydrocarbons are the aromatic organic molecules it is also known as arenes aromatic hydrocarbons have a special type of shape so it is the permanent member of benzene aromatic hydrocarbons is a very crucial topic of chemistry and the question of aromatic hydrocarbons related generally asked in many

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Sep 01 2020 web 24 the lowest temperature at which a hydrocarbon ignites is called its flash point using information in the table identify the correct statement a octane will ignite at 0 °c b hydrocarbons with the same molecular mass have the same flash point c the flash point of a hydrocarbon increases as the boiling point increases